3D RNA simulations

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Intricate structures

20-25	~30	40	60	80	100	3(00
microRNA siRNA	piRNA			tRNA riboswite	hes		
			ribozy	mes			
					snoRNA		
			Se la comparte de la			mR	NA
tRNA	telomera	se	viral frag	gment	(
riboswitch	n riboz	yme	triple he	lix			

Secondary structures classification



Canonical and non-canonical base pairing





288 theoretically possible pairs --> 145 found experimentally (NDB)

non-canonical base pairs



non-canonical base pairs



Exercice 1 Look at 3D structures

http://web.x3dna-dssr.org/ Look at PDB codes: 1Y26, 1L2X, 2K96

- For each system detect canonical pairs and non-canonical pairs.
- Are there multiple pairs (triplets or quadruplets)?
- How many stems are in the structure?
- What are the tertiary contacts?
- What is the secondary structure (topology) classification?

3D structure predictions

Homology modeling

Structures inferred by comparison with resolved structures of similar sequences

Reliable sequence alignment Extensive structural database





Exercice 2 Use online homology model server

http://iimcb.genesilico.pl/modernaserver/

- Launch test case 1QF6_B_tRNA.pdb
- Download 1QF6_B_tRNA.pdb on your machine and analyze it using DSSR.
- Download results from homology modeling and analyze them using DSSR.
- Detect the main similarities and differences between the two structures by comparing the DSSR results.
- Compare the 3D structures using a visualization software (Pymol, Chimera, VMD)

If you don't have homologues...

Hierarchical folding hypothesis



space to find 3D contacts

Secondary structure predictions

Genius prediction tool, Mfold, ViennaRNA

View Options Color: By Probability Show Bases Show Sequence Selection
Combinatorial problem when considering all possible base pair unknown energetic parameters
\sim .

- Based on a thermodynamic model considering the free energy of pair formation and stacking with the following pair : nearest-neighbor (Turner's) model
- Very fast
- Consider only canonical paris
- Does not include pseudoknots

3D structure predictions

Hybrids methods

Fragment reconstruction, 3D scaffolds, ... Good at predicting local structures Strongly rely on prediction of secondary structures first



Analyze sequence as short fragments (4-5 nt)

Predict 2D structure and assign each fragment to a specific 2D

Search a structural database for fragments of the same sequence and same 2D environment.

Assemble all fragments.

Relax (minimize) structure according to a given force field.

Exercice 3 Use online fragment reconstruction server http://rnacomposer.ibch.poznan.pl/

- Download the results on your machine and analyze it using DSSR.
- Download the results on your machine and analyze it using DSSR.
- What are the differences between the two 3D structures? (you can use a web server such as this: <u>http://rna.ucf.edu/WebSTAR3D</u> or your favorite visualization software)

Exercice 4 Use online scaffold modeling server http://rna.physics.missouri.edu/vfold3D/index.html

- Launch test case >example1 GCUCCUAGAAAGGCGCGGGGCCGAGGUACCAAGGCAGCGUGUGGAGC (((((....))))))))))
- Download the results on your machine and analyze it using DSSR.
- Compare the two prediction for this same system (RNAcomposer and Vfold)
- Launch a second test in which you modify the secondary structure given to the server (changes highlighted in red:

```
>example2
GCUCCUAGAAAGGCGCGGGGCCGAGGUACCAAGGCAGCGUGUGGAGC
(((((....)))))))))
```

- Download the results on your machine and analyze it using DSSR.
- Compare the two prediction for this same system (RNAcomposer and Vfold) (<u>http://rna.ucf.edu/</u> <u>WebSTAR3D</u>)

Limitations



VFold : 22.5 Å RMSD FARFAR : 11.6 Å RMSD

VFold : 3.7 Å RMSD FARFAR : 4.5 Å RMSD

2D helps, but it's not the end of the story

- canonical base pairing occurs in helices, but does not regulate long range interactions that ultimately determine the 3D structure (often non-canonical)
- pseudoknots are hard to account for.
- some molecules undergo BP shifts through their life that alter the 3D structure completely
- interactions that hold the molecule together are often non-canonical pairings

Physics-based simulations

Ab initio modeling

Physical models of the system, aiming at predicting equilibrium structures, folding intermediates, thermodynamics, kinetic barrier,...

Quantum : base-base interactions Atomistic : stability of known structures Coarse-grained : folding of a given sequence



RNA time scales and simulations



Simulation methods : 3D, potential energy functions

Molecular dynamics	serial	temporal evolution dynamics	limited sampling time	
F=ma energy model & force field	replica	enhanced sampling thermodynamics (replicas in T)	сри	
Monte Carlo simulations	simple	structure prediction	limited sampling time	
stochastic moves energy model	basin-hopping	energy landscape kinetics	сри	
stochastic moves energy model	basin-hopping cartesian coordinates	energy landscape kinetics small structural modifications	cpu limited applicability	

Simulation methods Classical molecular dynamics – force based

particles subject to force-field and friction (T dependent)



Simulation methods Monte Carlo – energy based



Quantum mechanics calculations

Ainima	Centers of hydro- gen bonds	QM	PM	AMBER BCC	AMBER RESP			
		m ⁹ Ade	(-10.6 ^a)					
1	N7, H62	-9.34	-10.84	-11.23	-11.08		M	
2	N1, H61	-8.39	-8.42	-9.40	-9.53	1	1161.1	
3	N3	-6.58	-6.97	-7.48	-8.08	7	51 > + 7	
		m ¹ Thy	(-10.4 ^a)			× -		
1	04	-4.62	-7.07	-8.59	-7.70			
2	O4, H3	-7.72	-8.85	-9.17	-8.97			
3	O2, H3	-7.39	-8.07	-8.90	-8.99			
4	02	-5.49	-6.89	-9.31	-8.55			
		m ⁹ Gua	(-14.0 ^b)					
1	N7, O6	-7.69	-12.41	-9.99	-10.44			
2	O6, H1	-10.42	-11.94	11.96	-12.59			
3	H1, H21	-9.02	-12.20					
4	N3, H22	-7.93	-9.86			65 -(3	65	
		m ¹ Cyt	(-11.4 ^a)	·		63	Se la construction de la constru	
1	H42	-5.42	-7.79			67	-	
2	N3, H41	-9.68	-11.22					
3	N3, O2	-6.81	-11.38		P 11			· 1
4	02	-6.07	-8.67		•//	R +-	- H	
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				5	201	U	V	A F
				A.	S S		3'	3
				- A	3	hairpin ₂₋₃	triplex-1	type-
				- 1	- and	NOC		
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Atomistic Force Fields

to compute both energies and forces for all kinds of simulations

Force fields depends on the representation used for the particles Typically referred to classical mechanics





CHARMM AMBER

... later

Vangaveti, Ranganathan, Chen, "Advances in RNA molecular dynamics: a simulator's guide to RNA force fields", Wiley Interdiscip Rev RNA. 2017

Šponer J, Bussi G, Krepl M, et al. RNA Structural Dynamics As Captured by Molecular Simulations: A Comprehensive Overview. Chem Rev. 2018

Atomistic force fields



Calibration against QM calculations

May include in the parametrization physical properties of pure model compounds (heat of vaporization, free energy of salvation, ...) data on conformational equilibria (NMR) and high resolution X-ray structural data.

! While protein folding is governed by hydrophobic effect and side chains packing, RNA folding is given by aromatic stacking, base-pair formation and RNA-ions interactions.

CHARMM

Parametrization philosophy:

- Complete scan of the QM potential energy surface (not just lowest-energy)
- Additional calculations at each H donor/acceptor site with the addition of a TIP3P water molecule -> indirect encoding of solvation properties
- Iterative optimization for best global agreement with input criteria (QM calculations + selected experimental structures and data such as various energies)



CHARMM + and -

Because of the highly delicate and iterative process it is hard to include new compounds such as ligands, modified nucleotides or chemical labels.

Recent publication of the parameters for the 100 naturally occurring modified RNAs

Transferability of the parameters from gas phase to solution phase is assumed (and not explicitly calibrated).

-> only TIP3P water molecules can be used (as they are the ones included in the original parametrization for solvation)

Use of high-level QM calculations give better description of relative energies between competing minima, hydrogen bonding energies and charge dipole moments than HF calculations that are commonly used to calibrate ff (less demanding). Dispersion-mediated interactions such as base stacking still require higher level of electron correlation to be accurately described.

-> Introduction of some degree of polarizability can help (Drude model).



AMBER

(Assisted Model Building and Energy Refinement)

AMBER ff94 : nucleic acids LJ parameters codeveloped together with other small organic compounds. Specific RNA torsions from fitting a QM (low-level) profile for the phosphate group and for the ribose sugar. Transitions between A and B-form helices.

AMBER ff99 : refit of all nucleotide torsions against detailed QM PES. AMBER ff99-BSCO : α/γ torsions -> eliminates backbone distorsions of B-form duplexes. AMBER ff99-BSC1 : ε/ζ torsions (sugar puckering) -> correctly reproduces twist and roll in DNA AMBER ff99- χ OL : refit of glycosidic torsion χ -> better anti/syn ratios, more realistic major groove widths.





AMBER



Parametrization protocol is linear, watermodel independent, and assumes transferability of parameters developed for small organic molecules

the modular and non-iterative nature of the parametrization scheme allows for improvements to be "easily" introduced and tested (that's why there are a zillions variants...)

Even though all water models can be used with AMBER (unlike CHARMM), simulation results are dependent on the model chosen.

AMBER example : tetraloop folding



Garcia 2008: REMD of 226 ns per replica, using 52 replicas.

-> unbiased folding of all replicas starting from extended conformations.



Bussi 2016: metadynamics + REMD -> native structure despite that it is not the most stable structure according to the ff

atomistic ff performance



- only 8-12 nt RNAs have been folded from random initial conformations (no matter the ff) and using enhanced sampling methods
- longer RNAs can be simulated from an experimental structure as long as large structural changes are not expected and/or being investigated.
- Due to the current difficulties in describing base-pairing energetics, atomistic simulations cannot be used to melt duplexes or observe conformational equilibria involving base-pairing rearrangements.

Current aa ff standing

Current RNA atomistic force fields are still inadequate

Differences in the ff arise from different optimization of torsions, but torsions have the LEAST physical basis of all the potentials in the ff. -> torsions are `the garbage' where we hide all the flaws in the ff

To assess the ff accuracy one simulates a variety of experimentally solved crystal structures for as long as is feasible to make sure that the molecule does not drift away.

-> RNA molecules are very flexible and can adopt various alternative conformations. Maintaining one single structure is not enough to ensure the correctness of the ff.

When QM is used for parametrization a transferability is assumed from properties in gas phase to properties in solution (not so obvious!) When ff contain calibrations from solution-phase experiments using empirical approaches, these perform best under those same conditions for calibration (transferability issues)

MD limitations



- At low temperatures the molecule gets trapped in local minima.
- At high temperature the molecule does not fold.

Configuration space not fully explored Enhanced sampling solves part of the problem

Ions – nucleic acids weak spot

At physiological conditions nucleic acids carry one -1 charge on each phosphate group



RNA and DNA are surrounded by positive metal ions!



Ions

Counter-ions (bulk)



Surface ions (ionic cloud)



Ion classification



Chelated: directly linked to RNA through coordination, immobile (ion clamps)

Glassy: closely associated with DNA and RNA, restricted mobility (i.e. ions in grooves)

Condensed: mobile ions in close distance from DNA or RNA, they explore the whole NA surface

Bulk: completely free ions, they explore the whole space, they provide the neutrality of the solution

Ion classification



- a) structure of P4-P6 domain of group 1 intron RNA
- b) envelope of condensed cations
- c) glassy cations in the grooves
- d) coordination of glassy Mg²⁺ ions
- e) highly coordinated structural Mg²⁺

Chelated ions

Structural ions are an integral part of the structure. They coordinate with surface water molecule and with phosphate groups to promote and stabilize RNA architectures



electrostatic as well as non-Coulombic interactions

polarization charge-transfer correlation

Modeling ions

Neutralizing the negative charge of the nucleic acids backbone

explicit ion simulations implicit representation

Monovalent ions : Na⁺, K⁺

Coulomb + LJ

Debye-Huckle

Divalent Mg²⁺

explicit water, explicit ion simulations : very slow water dynamics slow ionic diffusion -> poor statistics

implicit representation:
 how to account for coordination?

Polarization

Atomistic polarizable force-fields (Drude model)



Why ions pose problems

very delicate equilibrium with the environment



Ions (Mg+2 in particular) coordinate strongly with water.

Difficult to define LJ parameters for the interactions with ions and the other particles because the size changes if water is present or not.





Example : histones

The Nucleosome Core Particle (NCP):





Example : histones



Nucleosome Core Particle (NCP) model (L. Nordenskiold)



I = 16.6 mM (ionic strength)

Folding problem

Prediction of structure, dynamical and thermodynamical behavior in 3D



Accuracy and speed



What model?



Flat or round?



base-plane from 3 particles

Force field?

 $E = E_{\text{local}} + E_{\text{ex vol}} + E_{\text{BP}} + E_{\text{electrostatics}} + E_{\text{stacking}}$



geometric parameters
-> statistical analysis NDB

energetic parameters
-> experimental data, QM calculations, ?

Choosing functional forms

Long-range : Lennard Jones potential

Breakable bonds : Morse potential
$$V(x) = \epsilon \left(1 - e^{-a(x-x_0)}\right)^2$$

spherical particles $V(r, \theta, \phi) = V_1(r) \times V_2(\theta, \phi)$

Pure Coulomb
$$V(x) = \frac{q_1 q_2}{4\pi\varepsilon_0\varepsilon_r} \frac{1}{x}$$

Screened ionic potential: Debye-Huckel
$$V(x) = \frac{q_1 q_2}{4\pi\varepsilon_0\varepsilon_r} \frac{e^{-\frac{x}{\lambda}}}{x}$$

elliptical (anisotropic) particles: Guy-Berne potential

$$V(\vec{u}_i, \vec{u}_j, \vec{r}) = 4\epsilon(\vec{u}_i, \vec{u}_j, \vec{s}) \left\{ \left[\frac{\sigma_0}{4 - \sigma(\vec{u}_i, \vec{u}_j, \vec{s}) + \sigma_0} \right]^{12} - \left[\frac{\sigma_0}{4 - \sigma(\vec{u}_i, \vec{u}_j, \vec{s}) + \sigma_0} \right]^6 \right\}$$

Up or down?



Rely on the validity of the underlying model. Ok for proteins, but still problematic for nucleic acids.



Difficulty of extracting suitable information from experimental data.

What for?

NARES-2P (Sheraga 2015) : DNA melting, proof of principle on dipoles



TIS (Thirumalai 2012) : Ribozyme structural transitions on folding



What for?

Plotkin (2010) : DNA persistence length



OxDNA (Ouldridge 2010) : DNA origami and nanotechnology



What for?



Pseudoknot comparison example



Model's choices do matter



Same structural prediction. Impact on dynamics, thermodynamics



what's missing ?





Exercice 5 Look at an atomistic simulation

You are going to use the program VMD to visually analyze the trajectory of the molecule 1Y26

- Properly visualize the molecule.
 - Using different representations highlight the backbone of the molecule and the bases.
 - Based on the analysis of the initial structure by DSSR highlight stems (with different colors for example).
 - and identify the kissing loop.
- Visualize the trajectory and highlight the movements of the molecule.
 - Compute RMDS.
 - Superpose movements.
- Assess the stability of base pairs by plotting the time evolution of the distance between atoms forming hydrogen bonds. Can you detect differences between canonical and non-canonical pairs?

Exercice 6 Look at a CG simulation

You are going to use the program VMD to visually analyze the trajectory of the molecule 1Y26

- Visualize the molecule and familiarize with the coarse-grained representation. Notice that the visualization of VMD is not optimized for the GC model.
- Visualize the trajectory and highlight the movements of the molecule.
 - Compute RMDS.
 - Superpose movements.
- Assess the stability of base pairs by plotting the time evolution of the distance between atoms forming hydrogen bonds.
- Draw a comparison between atomistic and coarse-grained simulations for the system under investigation.